

EENV 444 Air Pollution Control II: Test # 2, Take Home Test: Fall 2024.

No group or consulting allowed.

Due on Dec 3, 2024, Tuesday in class 9:30 am or earlier.

Make appropriate assumptions when needed and justify

**1.0** A 1000 MWe coal fired powerplant burns 12,500 Btu/lb. coal containing 2.5% sulfur and 150 ppb of mercury. You are asked to determine the efficiency of control system required to meet the NSPS (Based on energy output) for SO<sub>2</sub> and Mercury. Also suggest the type of control systems you would recommend in order to meet the emission standards of SO<sub>2</sub> and Hg. The thermal efficiency of the powerplant is 38%. If the coal contains 60% carbon, what will be the amount of CO<sub>2</sub> emitted per day (tons) from this powerplant.

Following steps are used for grading your answers:

- a) Identify the Emission Standards for SO<sub>2</sub> and Hg ( based on energy output) (3)
- b) Determine the efficiency required to meet the above standards. Please show all your calculations (5)
- c) Justify the type of control system you have chosen to meet the standards. (2)
- d) Determine the amount of CO<sub>2</sub> emissions and a brief note on its effect on climate change (2)

**2.0** The company decided to use limestone FGD process to control the SO<sub>2</sub> emissions from the above power plant with 95% removal of sulfur dioxide. You are a new engineer and you are asked to determine the amount of solid sludge created by the limestone process. The limestone available is at 95% purity ( 5% inert) and the process needs 15% excess limestone. Also determine the amount of energy required to run the main fan and for pumping the sludge, given the following information: Every pound of coal burned produces 16 pounds of exhaust with a MW of 29 and the L/G ratio is 70 gallons per 1000 scf. The sludge has to be pumped with a 100 feet head loss to store it in the pond. The sludge has a specific gravity of 1.15. Fan efficiency is 0.7 and the fan pressure drop is 12 inches of H<sub>2</sub>O. The pump efficiency is 0.7.

Steps used to grade your answer are:

- a) identify equation to estimate the amount of limestone needed, theoretical and actual (4)
- b) identify the amount of waste stream , solids and liquids (4)
- c) Total Hp required for the fan, pump and heating the flue gas (4)

**3.0** A 2 feet diameter packed tower packed with 1-inch ceramic Raschig rings, with a gas to liquid mass flow ratio of 1, is treating a gas stream containing 1.5 % ammonia (volume basis). The outlet ammonia concentration shall not exceed 0.1 % by volume to meet the EPA requirements at the exit gas stream of the tower. The tower is working at 100 F and 1

atmosphere and the gas stream is at 100F and 1 atm. The height of transfer unit based on overall gas film coefficients is 1.85. The slope of the equilibrium line is 1.3

Steps used to grade your answer are:

- a) Determine the gas flow rate the packed tower could treat, to meet the emission requirements. (5)
- b) Determine the height of the packed tower. Show all calculations including the graph (5)

**4.0** A power plant is planning to use Wellman-Lord SO<sub>2</sub> control process. Determine the amount of makeup soda ash required per day for the coal fired power plant given the following information: 600 MWe plant, 38% thermal efficiency, 12,000 btu/lb. coal with 1.5 % sulfur, 95 % sulfur removal is required. Steps used to grade your answer are:

- a) determine the amount of SO<sub>2</sub> to be removed (3)
- b) determine the makeup soda ash required per day (3)

**5.0** a) Estimate the NO<sub>x</sub> emissions in ppm and kg/hr. from a coal containing 1.8% organic nitrogen combusted at an air fuel ratio of 1.5 of stoichiometric. One thousand kg of coal is combusted per hour and producing 22,000 kg/hr. of flue gas (MW=28). Ignore the thermal NO<sub>x</sub> for this problem. (5)

5 b) Calculate the stoichiometric amount of air required for complete combustion in kg of air/kg of fuel, of fuel oil represented by the formula C<sub>11</sub>H<sub>20</sub>N. Assume no Thermal or Fuel NO<sub>x</sub> is formed. Complete combustion products are CO<sub>2</sub>, H<sub>2</sub>O and NO

Solve problem 16.9 from our text book (3)

**6.0** Review "Code of Ethics" from two professional Societies and compare and contrast. The two professional societies ( you may consider any two of these): American Academy of Environmental Engineers, American Society of Civil Engineers, Air and Waste Management Association, American Chemical Society and others. (2)